SHORT COMMUNICATION

TRITERPENES AND STEROLS OF COFFEE OIL

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Abstract—Samples of triterpenes and sterols previously isolated from coffee oil and reported to contain lanosterol and coffeasterol have been reexamined by TLC and mass spectrometry. The presence of lanosterol is not confirmed; cycloartenol, cycloartanol and probably 24-methylenecycloartanol have been detected. The sterol fraction described as coffeasterol is a mixture of at least six methylsterols.

INTRODUCTION

In 1964, Kaufmann and Sen Gupta¹ analysed the unsaponifiable fraction of coffee oil. Three triterpenes were identified as lanosterol, dihydrolanosterol and squalene. A new C_{31} sterol was also found, coffeasterol, for which the structure 4-ethylidene 8,24-stigmastadiene- 3β -ol was proposed. In 1968, Goad² repeated the analysis of the triterpene alcohols in coffee oil and failed to find lanosterol. As authentic lanosterol has been rarely found in higher plants, its presence in coffee beans would have been of some interest.

TLC and mass spectrometry permitted a reexamination of the original preparations isolated in 1964, but the presence of lanosterol could not be confirmed.

RESULTS

Triterpene Alcohols

The TLC of the acetates showed three substances with R_f s of cycloartanol acetate (or dihydrolanosterol acetate; $ca.\ 10\%\ R_f\ 0.65$), cycloartenol acetate (or lanosterol acetate; $ca.\ 60\%\ R_f\ 0.45$) and 24-methylenecycloartanol (or 24-methylene dihydrolanosterol; $ca.\ 30\%\ R_f\ 0.35$). The three compounds were isolated by preparative TLC. The epoxide acetate of the main product ($R_f\ 0.45$) was prepared and compared by TLC with authentic epoxide acetates of lanosterol (diepoxide $R_f\ 0.20$) and of cycloartenol (monoepoxide $R_f\ 0.40$); it is thus identified as the epoxide of cycloartenol. Mass spectrometry of the acetate confirmed this result: molecular ion at $m/e\ 468$, 453 (M-15), 408 (M-60), 393 (M-60-15); a peak at $m/e\ 286$ is in agreement with the cyclopropane ring at 9–19.³ The acetate of $R_f\ 0.35$ possesses a molecular ion at 482 with main fragments at 467 (M-15), 422 (M-60), 407 (M-60-15); at $m/e\ 300$ is found a fragment corresponding to the presence of the cyclopropane ring at 9–19 and a side chain as in 24-methylenecycloartanol (the isomeric cyclolaudenol can not be

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¹ H. P. KAUFMANN and A. K. SEN GUPTA, Fette Seifen Anstrichmittel 66, 461 (1964).

² L. J. Goad, unpublished results cited by G. Ponsinet and G. Ourisson, *Phytochem*, 7, 762 (1968),

³ H. E. AUDIER, R. BEUGELMANS and B. C. DAS, Tetrahedron Letters 4341 (1966).

excluded). Too little of the product with R_f 0.65 was obtained for further study; but after epoxidation the acetate had an unchanged R_f ; this result and the R_f could fit cycloartanol.

Coffeasterol

TLC on SiO_2 -AgNO₃ showed two fractions of R_f 0.35 and 0.40 which were isolated and analysed by mass spectrometry. The substance of R_f 0.35 has a molecular ion at m/e 412 and probably a methylene group at C-24 as the "cyclic" elimination process of the side chain is observed with an ion at 328. The ion corresponding to the total elimination of the side chain is at 285 (M-C₉H₁₇ and transfer of 2H) and the fragmentation through cycle D leads to m/e 227. There are small peaks at M-18 and at M-15-18 in agreement with the lack of unsaturation in position 5. By deduction, the cyclic part of the molecule bears an extra methyl group, probably in position 4. From the nature of the different substitutions, the product could be 24-methylenelophenol or an isomer. The substance of R_f 0.40 is still a mixture of homologous series with molecular ions at m/e 428, 426, 414 and 400, the most abundant being at 426. The main component has a C_{10} side chain probably with an ethylidene group at C-24; ("cyclic" elimination of the side chain). This compound could be ethylidene-24-lophenol, and the minor homologues, with side chains in C_8H_{17} (m/e 400), C_9H_{19} (m/e 414), C_9H_{17} (m/e 412) and $C_{10}H_{21}$ (m/e 428) could consist of a biosynthetic series belonging to the lophenol family.

EXPERIMENTAL

The triterpene alcohols and the coffeasterol fractions re-investigated are those previously isolated by Kaufmann and Sen Gupta.¹ The acetates of the triterpene alcohols were separated by TLC on AgNO₃-impregnated alumina using hexane EtOAc (20:1). The coffeasterol fraction was chromatographed on SiO₂-AgNO₃ TLC with pentane-EtOAc-acetone (18:4:1). The epoxide acetates were prepared according to Ourisson *et al.*⁴ and separated on Al₂O₃-AgNO₃ with hexane-EtOAc (20:1). Mass spectrometric determinations were made on an Atlas CH₄ and an AEI MS9.*

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- * Mass spectrometric determinations were performed by MM. Cosson and Varenne under the direction of Dr. B. C. Das.
- ⁴ G. Ponsinet and G. Ourisson, Phytochem. 4, 799 (1965).